

4-Dimethylamino-6-(1-methyl-4-nitro-1*H*-pyrrol-2-yl)-1,3,5-triazine-2-amine dimethylformamide solvate

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## Key indicators

Single-crystal X-ray study

 $T = 173\text{ K}$ Mean  $\sigma(\text{C}-\text{C}) = 0.004\text{ \AA}$  $R$  factor = 0.067 $wR$  factor = 0.143

Data-to-parameter ratio = 13.1

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound,  $\text{C}_{10}\text{H}_{13}\text{N}_7\text{O}_2 \cdot \text{C}_3\text{H}_7\text{NO}$ , consisting of one molecule of 4-dimethylamino-6-(1-methyl-4-nitro-1*H*-pyrrol-2-yl)-1,3,5-triazine-2-amine and one dimethylformamide solvent molecule, is a member of a new series of 4,6-diaminotriazines. The triazine derivative molecule, containing a pyrrole and a triazine ring, is essentially planar, with all the non-H atoms being in the same plane;  $\pi$ -conjugation is observed. An extensive network of hydrogen bonds ( $\text{N}-\text{H} \cdots \text{N}$  and  $\text{N}-\text{H} \cdots \text{O}$ ) and  $\pi$ - $\pi$  stacking interactions maintain the crystal structure.

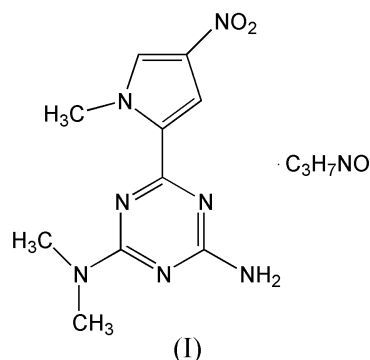
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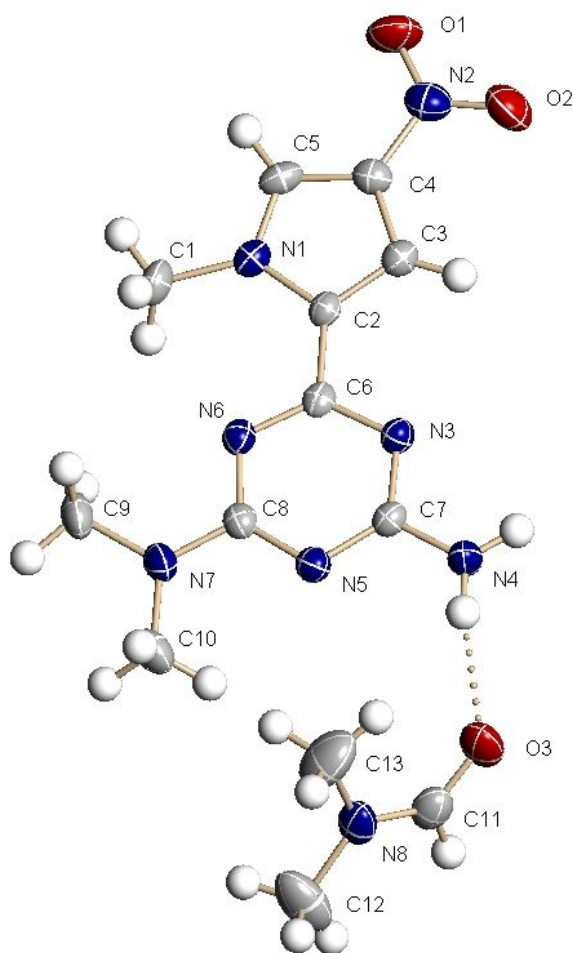
## Comment

Triazine derivatives have demonstrated a broad range of biological activities, including anti-angiogenesis, herbicidal effects, antimetastatic effects, Erm methyltransferase inhibition, antimicrobial effects (Bork *et al.*, 2003), the inhibition of the differentiation of endothelial progenitor cells (Park *et al.*, 2003) and the prevention of the early cell death of transplanted myogenic cells (El Fahime *et al.*, 2003). These biological functions of triazine derivatives stimulated our research interest, and we have synthesized the title triazine derivative, (I), from the conjugate of metformin and 1-methyl-4-nitropyrrole by cyclization of the biguanide group.



The main geometric parameters of (I) are listed in Table 1 and the molecular structure is illustrated in Fig. 1. Compound (I) contains a pyrrole and a triazine ring, which are connected by a C2–C6 bond of 1.467 (4) Å. In the molecule, these two rings are essentially coplanar and the least-squares plane containing all the non-H atoms has an r.m.s. deviation of only 0.0504 Å. Therefore, extensive  $\pi$ -conjugation exists between the two rings, encompassing also the two amine and one nitro groups.

The hydrogen-bond details are given in Table 2 and a packing diagram is shown in Fig. 2. Hydrogen bonds ( $\text{N}-\text{H} \cdots \text{N}$  and  $\text{N}-\text{H} \cdots \text{O}$ ) exist between the triazine derivatives



**Figure 1**  
The structure of (I), with displacement ellipsoids drawn at the 50% probability level.

and dimethylformamide solvent molecules. From the packing diagram, we found that the molecules of (I) are arranged in such a way that neighbouring molecules partly stack on top of each other, with a separation of about 3.40 Å, indicating significant  $\pi$ - $\pi$  stacking interactions (Zhang *et al.*, 2001).

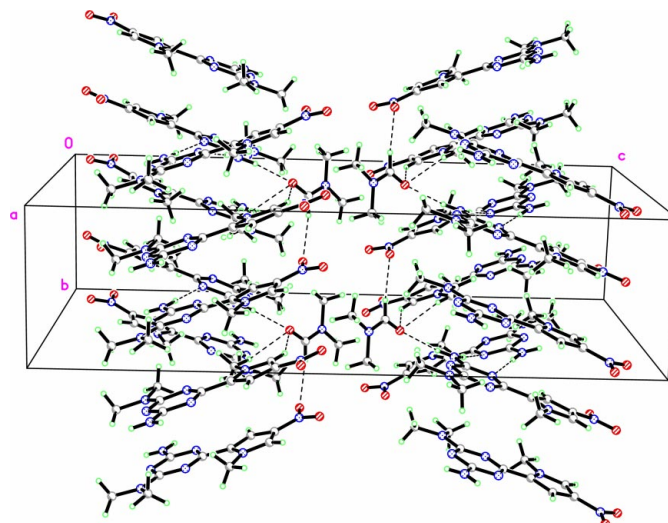
## Experimental

Crystals of (I) were grown from a hot dimethylformamide solution in which the title compound (1.0 mmol) was dissolved. The solution was left at room temperature and crystals formed in the bottom of the flask after one week. The elemental analyses were in agreement with the structural composition of (I).

### Crystal data

$C_{10}H_{13}N_7O_2 \cdot C_3H_7NO$   
 $M_r = 336.37$   
 Orthorhombic, *Pbcn*  
 $a = 15.810$  (3) Å  
 $b = 7.299$  (2) Å  
 $c = 28.550$  (6) Å  
 $V = 3294.6$  (12) Å<sup>3</sup>  
 $Z = 8$   
 $D_x = 1.356$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation  
 Cell parameters from 18 161 reflections  
 $\theta = 2.6$ – $24.5^\circ$   
 $\mu = 0.10$  mm<sup>-1</sup>  
 $T = 173$  (2) K  
 Pillar, colourless  
 $0.30 \times 0.30 \times 0.20$  mm



**Figure 2**  
The hydrogen-bonding network of (I), viewed approximately along the *a* axis.

### Data collection

Bruker SMART 1K CCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Sheldrick, 2000)  
 $T_{\min} = 0.970$ ,  $T_{\max} = 0.980$   
 14 666 measured reflections

2905 independent reflections  
 2217 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.055$   
 $\theta_{\max} = 25.0^\circ$   
 $h = -18 \rightarrow 18$   
 $k = -8 \rightarrow 4$   
 $l = -33 \rightarrow 33$

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.067$   
 $wR(F^2) = 0.143$   
 $S = 1.17$   
 2905 reflections  
 222 parameters

H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0562P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.19$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.17$  e Å<sup>-3</sup>

**Table 1**

Selected geometric parameters (Å, °).

C1–N1	1.476 (3)	C6–N3	1.340 (3)
C2–C3	1.364 (4)	C7–N4	1.338 (3)
C2–N1	1.397 (3)	C7–N5	1.338 (3)
C2–C6	1.467 (4)	C7–N3	1.355 (3)
C3–C4	1.393 (4)	C8–N5	1.340 (3)
C4–C5	1.375 (4)	C8–N7	1.347 (3)
C4–N2	1.422 (4)	C8–N6	1.356 (3)
C5–N1	1.341 (3)	C9–N7	1.461 (3)
C6–N6	1.324 (3)	C10–N7	1.451 (4)
C3–C2–N1	107.4 (2)	N5–C8–N6	125.3 (2)
C3–C2–C6	127.7 (2)	N7–C8–N6	116.5 (2)
N1–C2–C6	124.8 (2)	C5–N1–C2	109.3 (2)
C2–C3–C4	107.1 (2)	C5–N1–C1	121.9 (2)
C5–C4–C3	108.7 (2)	C2–N1–C1	128.8 (2)
C5–C4–N2	125.0 (3)	O1–N2–O2	123.3 (3)
C3–C4–N2	126.2 (3)	O1–N2–C4	118.8 (3)
N1–C5–C4	107.5 (2)	O2–N2–C4	117.9 (3)
N6–C6–N3	126.4 (2)	C6–N3–C7	113.6 (2)
N6–C6–C2	118.6 (2)	C7–N5–C8	114.3 (2)
N3–C6–C2	115.0 (2)	C6–N6–C8	114.5 (2)
N4–C7–N5	117.3 (2)	C8–N7–C10	121.3 (2)
N4–C7–N3	116.8 (2)	C8–N7–C9	120.9 (2)
N5–C7–N3	125.9 (2)	C10–N7–C9	117.4 (2)
N5–C8–N7	118.2 (2)		

**Table 2**  
Hydrogen-bonding geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N4-H4A \cdots O3$	0.88	2.12	2.987 (3)	170
$N4-H4B \cdots N3^i$	0.88	2.22	3.099 (3)	175

Symmetry code: (i)  $2-x, y, \frac{1}{2}-z$ .

H atoms attached to C and N atoms were placed in geometrically idealized positions, with  $Csp^3-H = 0.98 \text{ \AA}$ ,  $Csp^2-H = 0.95 \text{ \AA}$  and  $Nsp^2-H = 0.88 \text{ \AA}$ , and constrained to ride on their parent atoms, with  $U_{iso}(H) = 1.2U_{eq}(N \text{ and } Csp^2)$  and  $U_{iso}(H) = 1.5U_{eq}(Csp^3)$ .

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL/PC* (Sheldrick, 1999); software used to prepare material for publication: *SHELXTL/PC*.

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